

LETTERS
TO THE EDITOR

New Synthetic Approach to Phosphorylated 2-Aminothiazoles

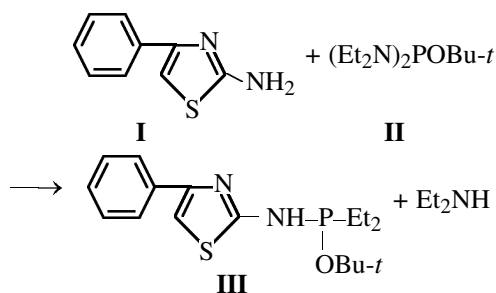
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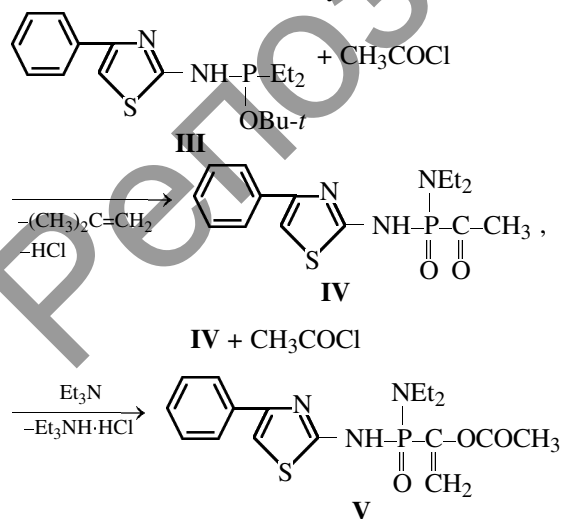
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It was shown previously that 2-amino-4-phenylthiazole (**I**) easily reacts with *tert*-butyl tetraethylphosphorodiamidite according to the transamidation scheme to form *tert*-butyl *N,N*-diethyl-*N'*-(4-phenylthiazol-2-yl)phosphorodiamidite (**III**) [1].



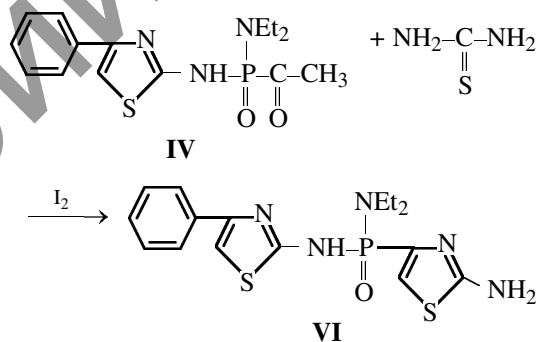
Compound **III** was reacted with acetyl chloride to find that the reaction proceeds regioselectively and results in formation of acetylphosphonic diamide **IV**. The structure of this product suggests existence of the enolic form, which was experimentally proved by the reaction with 2 mol of acetyl chloride.



Acetylphosphonic diamide **IV** presents interest as a promising starting material for synthesis of new

2-aminothiazole derivatives which are assuming increasing significance in pharmaceutical industry, biochemistry, and technics.

Prolonged heating of compound **IV** with thiourea in presence of crystalline iodine followed by conventional workup gave *N,N*-diethyl-*N'*-(4-phenylthiazol-2-yl)(2-aminothiazol-2-yl)phosphonic diamide (**VI**).



***N,N*-Diethyl-*N'*-(4-phenylthiazol-2-yl)(2-aminothiazol-2-yl)phosphonic diamide (**VI**)**. Yield 78%, mp 175°C. IR spectrum, ν , cm^{-1} : 419, 1518, 1623 (C_6H_5), 1443 ($\text{C}=\text{N}$), 3436 (NH), 3317 and 3285 (NH_2), 1198 ($\text{P}=\text{O}$). ^1H NMR spectrum, δ , ppm: 7.2 s (1H, thiazol C^5H), 7.40–7.48 m (5H, C_6H_5), 1.23 t (6H, CH_3 , $^3J_{\text{HH}}$ 7 Hz), 2.52 m (4H, CH_2N), 7.75 d (1H, NH , $^2J_{\text{PH}}$ 12 Hz). Found, %: C 48.45; H 5.42; N 17.78; P 7.69; S 16.98. $\text{C}_{16}\text{H}_{20}\text{N}_5\text{OPS}_2$. Calculated, %: C 48.84; H 5.12; N 17.80; P 7.87; S 16.30.

The IR spectra were recorded on a Specord IR-75 spectrometer in thin layer in the range 3700–400 cm^{-1} and on a Nicolet Avatar-360 device. The ^1H NMR spectra were taken on a Bruker DRX-500 (500 MHz) spectrometer against internal TMS.

REFERENCES

- Sal'keeva, L.K., Nurmaganbetova, M.T., and Minaeva, E.V., *Zh. Obshch. Khim.*, 2005, vol. 75, no. 12, p. 2065.